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Supporting Information

**A Novel Tungsten Coil Electrothermal Vaporizer with Composite Structure
Coupled with Dielectric Barrier Discharge Optical Emission Spectrometer for
Direct Determination of Trace Mercury**

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22 1. instrumentation

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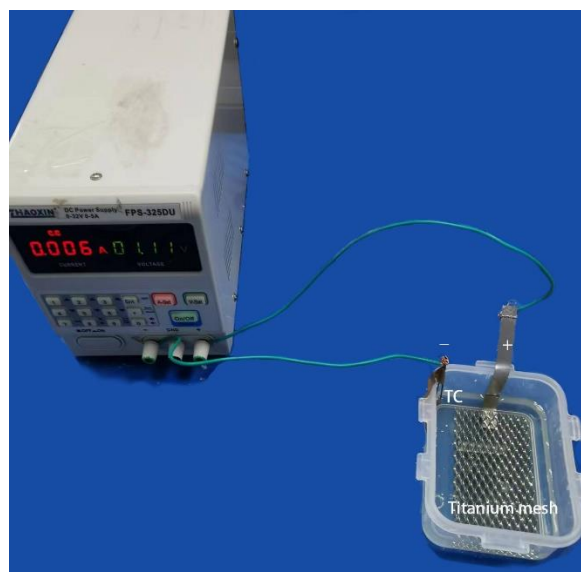


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25 **Fig S1.** Photograph of direct sampling Hg analyzer. This instrument with $0.57 \times 0.42 \times 0.19$ m dimensions, <12

26 kg weight, and ~100 W power consumption.

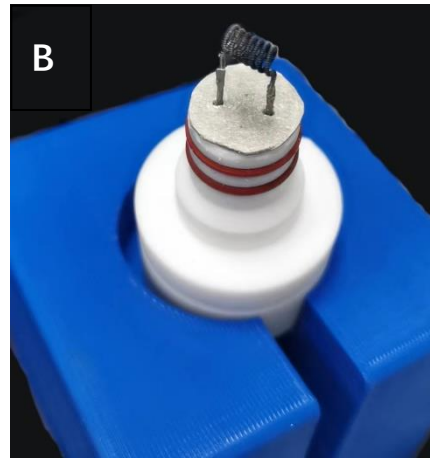
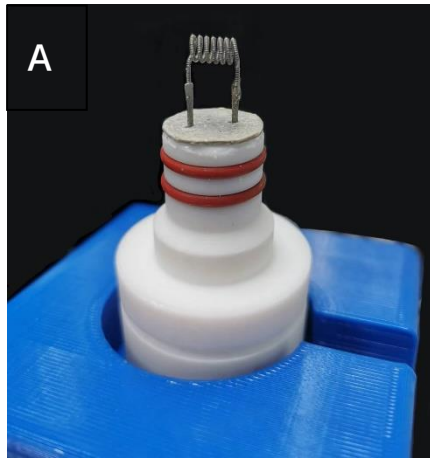
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29 **Fig S2.** Photograph of electrochemical plating instrument

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32 **Fig S3.** Photograph of platinumized TC used in different atmosphere. (A) platinumized TC used in Ar/H₂ mixture;
33 (B) TC used in Ar atmosphere that produced tungsten oxides (WO_x) (right).
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35 **2. Emission data handling**

36 Algorithm of emission data handling affects both quantitation precision and reproducibility. In ETV-DBD-
37 OES, it took certain time to completely release Hg. Signals can be calculated based on intensity, wavelength
38 and time, which are combined in 3 dimensions as peak volume. ¹Peak volume algorithm was reported to
39 significantly improve precision and linearity in electrothermal vaporization (ETV)-OES. Equations of peak
40 volume (Eq. 1,2) are shown below:

41
$$I_{area}(t) = \sum_{\lambda=\lambda_{start}}^{\lambda=\lambda_{end}} I_t(\lambda) \quad (1)$$

42
$$I_{volume} = \sum_{t=t_{start}}^{t=t_{end}} I_{area}(t) \quad (2)$$

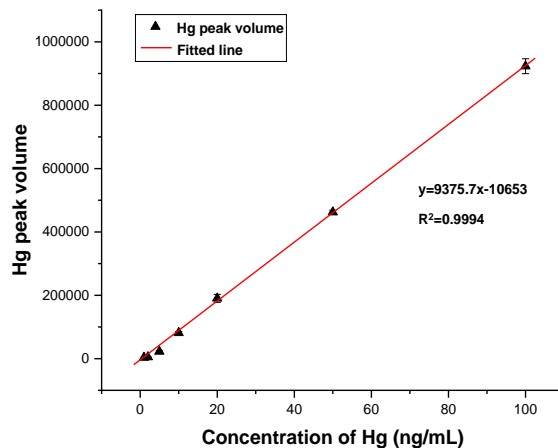
43 In Eq. 1, $I_{area}(t)$ is the calculated peak area at time t , $I_t(\lambda)$ is the intensity at time t , λ_{start} and λ_{end} are
44 the starting and ending wavelengths of the Hg peak. In Eq. 2, I_{volume} is the calculated peak volume, t_{start}
45 and t_{end} are the starting and ending time.

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48 **3. Standard curves**

49 The standard curves were constructed by plotting the calculated peak volume versus the analyte concentrations
50 of Hg solutions. The result was shown in Fig. S4. The linearity of R^2 was achieved to 0.9994. The LOD was
51 0.1 $\mu\text{g/L}$.



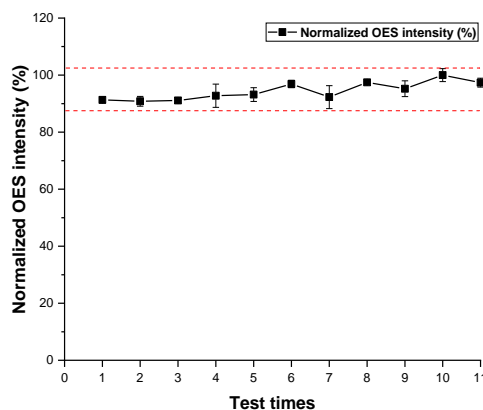
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Fig. S4. The calibration curve of Hg by this ETV-DBD-FOS. (n=3)

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55 **4. Stability**

56 To evaluate the stability of the instrument, the solution with $10\mu\text{g L}^{-1}$ Hg standard solution were continuously
57 analyzed for 11 times. The result was shown in Fig. S5. The RSD if the determination is 3.2%, which
58 demonstrated a good stability of this instrument.



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Fig. S5. Precision assessment of Hg solution. (n=3)

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62 1. J. Shan, W. Jin, B. Yu, Y. Ying, D. Zhu, H. Yu, Y. Yan and Q. Jin, *Journal of Chemometrics*,
63 2018, **32**.

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